Note

Synthesis of feruloylated and p-coumaroylated methyl glycosides

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There has been an expansion of the research effort directed at discerning the role of hydroxycinnamic acids in plant cell-wall development and degradation ¹⁻⁵. It is well documented that ferulic acid (4-hydroxy-3-methoxy-trans-cinnamic acid) and p-coumaric acid (4-hydroxy-trans-cinnamic acid) are esterified to the primary position of α -L-arabinofuranosyl units of arabinoxylans in plants of the Gramineae⁶⁻⁹. Ferulic acid has also been reported to be esterified to p-galactosyl units of spinach pectin¹⁰ and to the C-4 position of p-xylopyranosyl units of bamboo xyloglucans¹¹. Studies with grasses have shown that rumen microflora release free hydroxycinnamic acids from cell walls¹², while other investigations have indicated that free and combined acids can reduce the digestibility of forage cell-wall components^{13,14}. As hydroxycinnamic acids can also be covalently attached to lignin either through ester or ether linkages¹⁵⁻¹⁷, these difunctional molecules may provide a cross-link between lignin and polysaccharides which would ultimately influence both cell-wall development and degradation. Clearly, hydroxycinnamic acids contribute extensively to the chemical aspects of the plant cell wall.

We have recently turned our attention to determining the exact nature of the phenolic acid covalent interaction with lignin and polysaccharides and the influence of this interaction on plant cell-wall development and degradation. Our approach, which utilizes NMR spectroscopy, requires the synthesis and characterization of suitable model compounds and their use as substrates for isolated enzymes and rumen microbes. We desired a facile route to methyl glycosides with hydroxycinnamic acids esterified to the primary position. This note describes two

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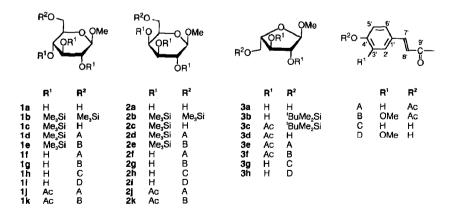
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simple, high-yielding routes to several appropriately substituted glycosides. The methods are quite amenable to the synthesis of ¹³C-labeled feruloylated and p-coumaroylated pyranosides and furanosides.

RESULTS AND DISCUSSION

The primary positions of methyl β -D-glucopyranoside (1a) and methyl β -D-galactopyranoside (2a) were readily available for acylation by the selective O-desilylation of 1b and 2b with methanolic potassium carbonate as has been described 19,20. Although the persilylated ethers were susceptible to degradation, 1c and 2c were quite stable (unless kept in solution for extended periods without adequate protection from moisture). The subsequent reaction of 1c and 2c with the 4-acetates of p-coumaroyl and feruloyl chlorides in pyridine afforded the crude silylated hydroxycinnamoyl carbohydrate esters (1d and e, and 2d and e, respectively) which were O-desilylated by exposure to aqueous ethanol. This route provided the crystalline 4'-acetoxycinnamoyl pyranosides in good yields. The D-galactopyranosides 2f and 2g were prepared in greater than 80% overall yield based on 2a. The overall yields of the D-glucopyranosides 1f and 1g were 45–50%, which was due to the susceptibility of 1b to nonselective O-desilylation. Removal of the 4'-acetate to provide model compounds 1h and i, and 2h and i was achieved by treatment with 0.5 M piperidine or pyrrolidine in 95% ethanol.

Removal of the trimethylsilyl group attached to the C-5 position of methyl 2,3,5-tri-O-trimethylsilyl-α-L-arabinofuranoside with K₂CO₃-methanol did not occur with any selectivity. Thus a different reaction scheme was developed to provide the hydroxycinnamic acid esters of **3a**. We have previously described an extremely simple synthesis of **3h** by taking advantage of the higher selectivity of the primary hydroxy group towards acylation²¹. However, the low yields (56%) were not suitable when ¹³C-labeled hydroxycinnamoyl chlorides were used in the coupling step.



The strategy employed utilizes the synthesis of 3b, the O-(tert-butyldimethylsilyl) derivative (TBDMS), which was available in 89% yield from 3a with tert-butyldimethylsilyl chloride-pyridine. Protection of the remaining hydroxyl groups could be accomplished in the same reaction flask by the subsequent addition of acetic anhydride. Purification by silica gel chromatography gave 3c in 84% yield from 3a. Acetylation was the protection method of choice due to the simplicity of the one-pot silylation-acetylation sequence and the need for peracetylated materials for comparison with acetylated native cell-wall tissue isolates. Acetylation of isolated cell-wall fractions is a standard approach to improve native material solubility, which permits the solution NMR analysis of more concentrated samples.

Cleavage of the tert-butyldimethylsilyl group to expose the 5-hydroxyl group was accomplished in 83% yield by treatment with 80% acetic acid for 24 h. The crude product obtained from the one-pot silylation-acetylation sequence can also be submitted to the O-desilylation step, thus avoiding an intermediate chromatographic purification. Subsequent silica gel chromatography affords 3d in greater than 70% overall yield from 3a. This procedure is an improvement over previous preparations of methyl 2,3-di-O-acetyl-L- and D-arabinofuranosides^{22,23}, which relied upon formation of the 5-O-trityl derivative. Coupling of 3d with the protected hydroxycinnamoyl chlorides in pyridine resulted in the formation of the peracetylated esters 3e and 3f in yields of up to 97%. Deacetylation of the fully protected L-arabinofuranoside esters was attempted with NaOMe-methanol (Zemplén method), piperidine-95% ethanol (50°), and pyrrolidine-95% ethanol (room temperature). The Zemplén deacetylation procedure resulted in 55-65% yield of the esters with the major byproducts being 1a and the methyl hydroxycinnamate. The piperidine treatment resulted in about the same overall yield (60-65%), whereas the pyrrolidine treatment (0.5 M, 24 h) provided the deprotected models 3g and 3h in 73-83% yields. The ability to remove acetate protecting groups with piperidine or pyrrolidine in 95% ethanol without affecting the hydroxycinnamate bond is quite convenient. A cinnamic acid ester provides a relatively strong bond compared to that of an acetate²⁴, thus allowing differentiation of the two acyl groups. The use of cinnamovl chloride in carbohydrate synthetic schemes warrants further study, especially in conjunction with standard acetate protection strategies.

The unambiguously assigned 13 C chemical shifts for several compounds are presented in Table I and are based on heteronuclear shift-correlated (HETCOR) spectroscopic experiments, with long-range 13 C- 1 H correlation experiments used to assign the feruloyl-containing compounds. It is apparent that the aromatic sidechain carbons (C-7', 8', and 9') are relatively insensitive to the carbohydrate and aryl moietics, as well as to acetylation. The C-7' carbon resonances range from 144.8 to 146.1 ppm for samples in acetone- d_6 . The C-8' carbon resonances behave in a similar fashion, although there is a significant difference in the frequencies of the acetylated vs. free compounds (118.9–118.2 vs. 115.5–115.2 ppm, respectively). The unprotected feruloylated glycosides 1i, 2i, and 3h exhibited C-8' resonances upfield of C-5' (115.5 vs. 116.0 ppm in acetone- d_6 ; 114.9 vs 115.9 ppm in 9:1

TABLE I $$^{13}{\rm C-NMR}$$ chemical shifts for selected compounds a,b

Carbon c	Chemic	Chemical shifts (ppm)	(mdd													
	p-Gluc	o-Glucopyranosyl	1				D-Galac	-Galactopyranosyl	syl				L-Arabin	-Arabinofuranosy	l⁄s	
	Coumaroyl	royl		Feruloy	-		Coumaroyl	royl		Feruloy			Coumaroyl	oyl	Feruloy	
	*1	1 h	3	*	ij	1k	2f *	2h	23	2g *	2i	2k	3e	38	3£	3h
1	103.9	104.8	102.0	103.9	104.8	102.0	104.4	105.2	102.5	104.4	105.3	102.4	107.5	110.2	107.5	110.3
2	73.6	74.5	72.0	73.6	74.6	72.0	73.3	72.0	9.69	73.3	72.0	69.5	82.0	83.1	82.0	83.2
3	76.4	9.77	73.5	76.4	77.7	73.6	70.9	74.4	71.7	71.0	74.4	71.7	78.0	79.2	78.1	79.3
4	70.4	71.2	9.69	70.3	71.2	69.5	69.1	69.7	68.4	69.1	9.69	68.3	81.2	82.3	81.3	82.4
S	73.9	74.7	72.3	73.9	74.8	72.3	72.9	73.3	71.4	72.9	73.3	71.4	64.0	64.7	63.9	64.8
9	64.1	64.2	63.0	64.0	64.2	67.9	64.3	64.0	62.3	64.3	64.0	62.3	ı	1	ı	I
1,	132.3	126.7	132.8	133.7	127.3	134.1	132.2	126.6	132.7	133.7	127.3	133.9	132.7	126.8	134.0	127.4
2,	130.0	130.9	130.2	112.2	111.2	112.3	130.0	130.9	130.2	112.2	111.3	112.2	130.1	130.9	112.3	111.2
3,	122.8	116.6	123.2	151.6	148.7	152.7	122.8	116.7	123.2	151.7	148.7	152.5	123.2	116.6	152.5	148.7
4,	152.7	160.6	153.5	141.7	150.0	142.7	152.6	160.5	153.5	141.7	150.1	142.6	153.4	160.5	142.6	150.1
5,	122.8	116.6	123.2	123.7	116.0	124.1	122.8	116.7	123.2	123.7	116.0	124.0	123.2	116.6	124.1	116.0
,9	130.0	130.9	130.2	121.9	124.0	122.4	130.0	130.9	130.2	122.0	124.0	122.3	130.1	130.9	122.2	124.1
7,	144.9	145.6	144.9	145.2	145.9	145.3	145.0	145.6	145.0	145.3	146.0	145.4	144.8	145.7	145.2	146.1
,8	117.9	115.1	118.5	118.1	115.4	118.5	117.8	115.2	118.3	118.1	115.5	118.2	118.5	115.2	118.6	115.5
9,	168.0	167.6	166.6	168.0	167.6	166.7	167.9	167.4	166.4	168.0	167.4	166.5	166.6	167.4	166.7	167.4
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^a In acetone-d₆; asterisk (*) indicates 1:1 (v/v) D₂O-acetone-d₆. ^b Chemical shifts for the acetate and methoxyl carbons are in the Experimental section. ^c For numbering sequence, refer to formulas.

acetone- d_6 : D_2O) which confirms the work of Ishii and Hiroi¹¹ who found that the assignments given for these two carbons in previous studies were interchanged.

EXPERIMENTAL*

General.—Melting points are uncorrected. Evaporations were performed under reduced pressure at temperatures not exceeding 42° (unless otherwise stated). Optical rotations were obtained at 20° with a Perkin-Elmer model 141 polarimeter. NMR spectra were initially recorded with a Bruker AM-360 360 MHz wide-bore instrument fitted with a 5-mm probe and operated at 24-27°; and subsequently with a Bruker AMX-360 spectrometer. One- and two-dimensional NMR spectra were obtained using standard pulse-sequence programs. Chemical shifts (ppm) are relative to the central solvent peaks of acetone- d_6 (δ_C 29.8, δ_H 2.04 ppm), Me₂SO- d_6 (δ_C 39.5, δ_H 2.49 ppm), or CDCl₃ (δ_C 77.0, δ_H 7.24 ppm).

Thin-layer chromatography was performed with Alugram Sil-G/UV₂₅₄ plates (Macherey-Nagel) with visualization either with UV light or by charring (5% $\rm H_2SO_4$ in 95% EtOH). Preparative TLC was carried out with Kieselgel-60 on pre-coated plates (E. Merck) and column chromatography was with Kieselgel 60 (E. Merck) using a standard flash chromatography apparatus (Ace Glass). Solvent systems employed were as follows: A, 50:1 $\rm CCl_4$ -acetone; B, $\rm CHCl_3$; C, 19:1 $\rm CHCl_3$ -EtOAc; D, 9:1 $\rm CHCl_3$ -EtOAc; D, EtOAc-HOAc; D, 15:1 $\rm EtOAc$ -MeOH.

4-Acetoxycinnamoyl chlorides.—4-O-Acetylferulic acid and 4-O-acetylferuloyl chloride were prepared as previously reported 21 . p-Coumaric acid (25 g, 152 mmol) was acetylated in pyridine (45 mL) with acetic anhydride (40 mL, 424 mmol). The mixture was left for 4 h and quenched by pouring into ice—water (1 L) with stirring. A white precipitate that fell out during mixing was isolated by filtration, washed with $\rm H_2O$, and air-dried. Crystallization from MeOH afforded 4-O-acetyl-p-coumaric acid (25.5 g, 81%): mp 205–211°; NMR (Me₂SO- d_6): $\delta_{\rm H}$ 2.25 (Ac), 6.47 (d, 1 H, $J_{7,8}$ 16.0 Hz, H-8), 7.16 (m, 2 H, J 8.6 Hz, H-3,5), 7.58 (d, 1 H, H-7), 7.71 (m, 2 H, H-2,6); $\delta_{\rm C}$ 20.9 (Ac), 119.4 (C-8), 122.4 (C-3,5), 129.5 (C-2,6), 132.1 (C-1), 143.1 (C-7), 152.0 (C-4), 167.7 (C-9), 169.1 (C=O).

The acid chlorides were prepared by refluxing a mixture of the 4-O-acetoxycin-namic acid (25 mmol) and thionyl chloride (4.5 mL, 3.8 equiv) in benzene (50 mL) for 45 min. The resulting clear solutions were evaporated to a solid, redissolved in toluene, and evaporated to a solid again. Crystallization from hot toluene gave the 4-acetoxycinnamoyl chlorides in 75–80% yields. 4-O-Acetyl-p-coumaroyl chloride: mp 118.5–121.5°; NMR (acetone- d_6): $\delta_{\rm H}$ 2.28 (Ac), 6.87 (d, 1 H, $J_{7.8}$ 15.6 Hz, H-8),

^{*} Mention of a trade name, proprietary product, or specific equipment does not constitute a guarantee of the product by the USDA and does not imply its approval to the exclusion of other products that may also be suitable.

7.26 (m, 2 H, *J* 8.7 Hz, H-3,5), 7.87 (m, 2 H, H-2,6), 7.95 (d, 1 H, H-7); $\delta_{\rm C}$ 21.0 (Ac), 122.8 (C-8), 123.4 (C-3,5), 131.5 (C-1,2,6), 150.9 (C-7), 154.7 (C-4), 166.1 (C-9), 169.2 (C=O).

Methyl 2,3,4-tri-O-trimethylsilyl-β-D-galactopyranoside (2c).—Methyl β-D-galactopyranoside (2a, 2.50 g, 12.9 mmol) was silylated with trimethylsilyl chloride (8 mL, 4.9 equiv) in pyridine (25 mL) for 5.5 h²⁵. The mixture was diluted with toluene and evaporated to a syrupy solid. Addition and evaporation of toluene was repeated twice, and the resulting product was diluted with petroleum ether and filtered. The filtrate was evaporated to a syrup and placed under high vacuum (150 mtorr) for 1 h to provide 6.35 g of crude 2b. The syrupy material was diluted with MeOH (100 mL), and to this solution was added (with stirring) methanolic K₂CO₃ (3 mL, 4.48 mg/mL)¹⁸. The mixture was left for 90 min at which time TLC indicated complete disappearance of the starting material. The solution was neutralized with HOAc (5 mL, 2.8 mg/mL MeOH) and evaporated to a syrup. Purification with silica gel chromatography (40 g, solvent D) provided 2c (4.88 g, 92% based on 2a) which crystallized from the syrup: mp $128.5-130^{\circ}$ [α]_D -8.2° (c 1.97, acetone); lit.¹⁹ mp 130.5–132°, $[\alpha]_D$ –1.72° (c 10.25, CHCl₃); NMR (CDCl₃): $\delta_{\rm H}$ 0.095 and 0.13 (SiCH₃), 3.39 (dd, 1 H, $J_{3,4}$ 2.7 Hz, $J_{3,2}$ 9.3 Hz, H-3), 3.45 (ddd, 1 H, $J_{5,4}$ 0.9 Hz, $J_{5,6a}$ 4.5 Hz, $J_{5,6b}$ 7.4 Hz, H-5), 3.48 (s, 3 H, OCH₃), 3.57–3.64 (m, 1 H, H-6a), 3.61 (dd, 1 H, J_{2.1} 7.5 Hz, H-2), 3.74 (dd, 1 H, H-4), 3.83 (ddd, 1 H, J 2.6 Hz, J_{eem} 11.2 Hz, H-6b), 4.08 (d, 1 H, H-1); $\delta_{\rm C}$ 0.39, 0.57, and 0.61 (SiCH₃), 57.2 (OCH₃) 62.7 (C-6), 72.0 (C-2), 72.3 (C-4), 75.0 (C-5), 75.1 (C-3), 105.1 (C-1).

Methyl 2,3,4-tri-O-trimethylsilyl-β-D-glucopyranoside (1c).—Compound 1c was prepared as 2c, with the yield being significantly lower (62% from 1a): mp 72–73°; $[\alpha]_D$ –6.0° (c 1.4, acetone); lit.²⁰ mp 72°, $[\alpha]_D$ –8.6° (c 0.5, CHCl₃); NMR (CDCl₃): δ_H 0.95, 0.11 and 0.12 (SiCH₃), 3.23 (dd, 1 H, $J_{2,1}$ 7.6 Hz, $J_{2,3}$ 8.8 Hz, H-2), 3.19–3.25 (m, 1 H, H-5), 3.39–3.42 (m, 2 H, H-3,4), 3.44 (OCH₃), 3.62 (ddd, 1 H, $J_{2,1}$ 5.3 and 7.1 Hz, J_{gem} 11.7 Hz, H-6a), 3.77 (ddd, 1 H, $J_{2,1}$ 2.8 and 6.3 Hz, H-6b), 4.08 (d, 1 H, H-1); δ_C 0.82, 1.01 and 1.29 (SiCH₃), 56.8 (OCH₃), 62.0 (C-6), 71.5 (C-4), 76.0 (C-2,5), 78.3 (C-3), 104.4 (C-1).

Methyl 6-O-(4'-O-acetylferuloyl)-β-D-galactopyranoside (2g).—Compound 2c (2.03 g, 4.94 mmol) was dissolved in pyridine (25 mL), and 4-O-acetylferuloyl chloride (1.37 g, 1.09 equiv) was added with stirring. After 5 h the crude silylated reaction product was diluted with toluene and evaporated to a syrupy solid. The mixture was diluted with petroleum ether and filtered. The filtrate was evaporated to a syrup and diluted with 75% aq EtOH (30 mL) and left unstirred in the dark until O-desilylation was complete (3 days). Processing and subsequent crystallization from EtOH gave 2g as white needles (1.85 g, 91%): mp 194.5–195.5°; [α]_D 9.4° (c 0.98, acetone–H₂O); NMR [1:1 (v/v) D₂O-acetone-d₆]: $\delta_{\rm H}$ 2.19 (Ac), 3.41 (OCH₃), 3.46 (dd, 1 H, $J_{2,1}$ 7.7 Hz, $J_{2,3}$ 9.8 Hz, H-2), 3.54 (dd, 1 H, $J_{3,4}$ 3.4 Hz, H-3), 3.75 (OCH₃), 3.78–3.82 (m, 1 H, H-5), 3.88 (dd, 1 H, $J_{4,5}$ 1.1 Hz, H-4), 4.18 (d, 1 H, H-1), 4.27 (dd, 1 H, $J_{6a,5}$ 4.6 Hz, J_{gem} 11.6 Hz, H-6a), 4.34 (dd, 1 H, $J_{6b,5}$ 7.7 Hz, H-6b), 6.46 (d, 1 H, $J_{8',7'}$ 16.0 Hz, H-8'), 7.02 (d, 1 H, $J_{5',6'}$ 8.1 Hz, H-5'),

7.14 (dd, 1 H, $J_{6',2'}$ 1.8 Hz, H-6'), 7.25 (d, 1 H, H-2'), 7.57 (d, 1 H, H-7'); $\delta_{\rm C}$ 20.2 (OAc); 56.2, 57.0 (OCH₃).

Anal. Calcd for C₁₉H₂₄O₁₀: C, 55.34; H, 5.87. Found: C, 55.11; H, 5.92.

Methyl 6-O-(4'-O-acetyl-p-coumaroyl)-β-D-galactopyranoside (2f).—Compound 2f was prepared according to the procedure for 2g with the use of 4-O-acetyl-p-coumaroyl chloride to afford 2f in 90% yield, which crystallized from absolute EtOH as white needles: mp 125–132°; [α]_D 9.8° (c 1.02, acetone–H₂O); NMR [1:1 (v/v) D₂O-acetone-d₆]: $\delta_{\rm H}$ 2.19 (Ac), 3.41 (OCH₃), 3.46 (dd, 1 H, $J_{2,1}$ 7.6 Hz, $J_{2,3}$ 9.8 Hz, H-2), 3.55 (dd, 1 H, $J_{3,4}$ 3.4 Hz, H-3), 3.78–3.82 (m, 1 H, H-5), 3.88 (dd, 1 H, $J_{4,5}$ 1.1 Hz, H-4), 4.18 (d, 1 H, H-1), 4.27 (dd, 1 H, $J_{6a,5}$ 4.6 Hz, J_{gem} 11.6 Hz, H-6a), 4.33 (dd, 1 H, $J_{6b,5}$ 7.7 Hz, H-6b), 6.42 (d, 1 H, $J_{8',7'}$ 16.0 Hz, H-8'), 7.04–7.07 (m, 2 H, H-3',5'), 7.53–7.60 (m, 3 H, H-2', 6', 7'); $\delta_{\rm C}$ 20.7 (Ac); 57.0 (OCH₃).

Anal. Calcd for C₁₈H₂₂O₉: C, 56.54; H 5.80. Found: C, 56.58; H, 5.76.

Methyl6-O-(4'-O-acetyl p-coumaroyl)-β-D-glucopyranoside (1f).—Coupling procedures were as for 2g. Purification by silica gel chromatography (40 g, solvent E, 200 mL), followed by solvent G (9200 mL)), and subsequent crystallization gave 1f (82%): mp 160–164°; $[\alpha]_D$ – 14.5° (c 1.2, acetone–H₂O); NMR [1:1 (v/v) D₂O-acetone- d_6]: δ_H 2.20 (Ac), 3.18 (dd, 1 H, $J_{2,1}$ 8.0 Hz, $J_{2,3}$ 9.1 Hz, H-2), 3.35–3.45 (m, 5 H, OCH₃, H-3, 4), 3.54–3.59 (m, 1 H, H-5), 4.24 (d, 1 H, H-1), 4.28 (dd, 1 H, $J_{6a,5}$ 5.9 Hz, J_{gem} 12.1 Hz, H-6a), 4.43–4.46 (m, 2.5 H, $J_{6b,5}$ 2.2 Hz, HDO and H-6b), 6.44 (d, 1 H, $J_{8',7'}$ 16.1 Hz, H-8'), 7.05–7.09 (m, 2 H, H-3',5'), 7.56–7.61 (m, 3 H, H-2', 6',); δ_C 20.7 (Ac); 57.0 (OCH₃).

Anal. Calcd for C₁₉H₂₄O₁₀: C, 56.54; H, 5.80. Found: C, 56.57; H, 5.76.

Methyl 6-O-(4'-O-acetylferuloyl)-β-D-glucopyranoside (1g).—The same procedure employed for 2g was used. Purification with silica gel chromatography (40 g, solvent E) and subsequent crystallization from 95% EtOH gave 1g (83%) as small hygroscopic white needles: mp 127.5–129.5°; $[\alpha]_D$ – 16.6° (c 0.92, acetone–H₂O); NMR [1:1 (v/v) D₂O-acetone- d_6]: δ_H 2.19 (Ac), 3.18 (dd, 1 H, $J_{2,1}$ 8.0 Hz, $J_{2,3}$ 9.1 Hz, H-2), 3.34–3.45 (m, 5 H, H-3, 4, OCH₃), 3.54–3.59 (m, 1 H, H-5), 3.75 (OCH₃), 4.24 (d, 1 H, H-1), 4.28 (dd, 1 H, $J_{6a,5}$ 5.8 Hz, J_{gem} 12.1 Hz, H-6a), 4.44 (dd, 1 H, $J_{6b',5'}$ 2.2 Hz, H-6b), 6.48 (d, 1 H, $J_{8',7'}$ 16.0 Hz, H-8'), 7.02 (d, 1 H, $J_{5',6'}$ 8.2 Hz, H-5'), 7.14 (dd, 1 H, $J_{6',2'}$ 1.9 Hz, H-6'), 7.26 (d, 1 H, H-2'), 7.57 (d, 1 H, H-7'); δ_C 20.2 (Ac) 56.2, 57.0 (OCH₃).

Anal. Calcd for C₁₇H₂₂O₉: C, 55.34; H, 5.87. Found: C, 55.21; H, 5.93.

Deacetylations.—The acetylated derivative (500 mg) was suspended in 95% EtOH (5 mL), and pyrrolidine (250 μ L) was added (which caused the solution to turn yellow). The starting material typically dissolved within 15 min and the reaction was allowed to continue for a total of 1 h. The mixture was added directly to a column of strongly acidic ion-exchange resin [Amberlite IRA-120 (H⁺), 40 mL, washed and packed in 95% EtOH]. Collection of the appropriate fractions provided the deprotected esters.

Methyl 6-O-(p-coumaroyl)- β -D-galactopyranoside (2h).—The material obtained

from the ion-exchange column could be crystallized from hot $\rm H_2O$ to give **2h** as off-white needles (407 mg, 90%): mp 103.5–106.5; $[\alpha]_D$ 14.7° (c 1.09, acetone– $\rm H_2O$); NMR (9:1 acetone- $\rm d_6$ – $\rm D_2O$): δ_H 3.43 (OCH₃); 3.50–3.59 (m, 2 H, H-2, 3); 3.79 (ddd, 1 H, $\rm J_{5,4}$ 1.2 Hz, $\rm J_{5,6a}$ 4.9 Hz, $\rm J_{5,6b}$ 7.4 Hz, H-5); 3.90 (dd, 1 H, $\rm J_{4,3}$ 3.1 Hz, H-4); 4.18 (d, 1 H, $\rm J_{1,2}$ 7.5 Hz, H-1); 4.29 (dd, 1 H, $\rm J_{gem}$ 11.5 Hz, H-6a); 4.35 (dd, 1 H, H-6b) 6.30 (d, 1 H, $\rm J_{8',7'}$ 16.0 Hz, H-8'); 6.83 (m, 2 H, H-3', 5'); 7.45 (m, 2 H, H-2', 6'); 7.57 (d, 1 H, H-7'); δ_C (acetone- $\rm d_6$), 56.6 (OCH₃).

Anal. Calcd for $C_{16}H_{20}O_8$: C, 56.47; H 5.92. Found: C, 56.39; H, 5.95.

Methyl 6-O-feruloyl-β-D-galactopyranoside (2i).—The fractions collected from the ion-exchange column were processed to afford 2i as a white foam (461 mg, 95%): $[\alpha]_D$ 14.1° (c 0.75, acetone– H_2O); NMR (9:1 acetone- d_6 – D_2O), δ_H 3.43 (OCH₃); 3.48–3.59 (m, 2 H, H-2, 3); 3.79 (m, 1 H, H-5); 3.85 (OCH₃); 3.90 (dd, 1 H, $J_{4,5}$ 1.2 Hz, $J_{4,3}$ 3.0 Hz, H-4); 4.18 (d, 1 H, $J_{1,2}$ 7.5 Hz, H-1); 4.31 (dd, 1 H, $J_{6a,5}$ 5.0 Hz, J_{gem} 11.4 Hz, H-6a); 4.37 dd, 1 H, $J_{6b,5}$ 7.3 Hz, H-6b); 6.37 (d, 1 H, $J_{8',7'}$ 15.9 Hz, H-8'); 6.84 (d, 1 H, $J_{5',6'}$ 8.2 Hz, H-5'); 7.08 (dd, 1 H, $J_{2',6'}$ 1.9 Hz, H-6'); 7.24 (d, 1 H, H-2'); 7.58 (d, 1 H, H-7'); δ_C (acetone- d_6), 56.3, 56.6 (OCH₃).

Anal. Calcd for $C_{17}H_{22}O_9$: C, 55.13; H, 5.99. Found: C, 54.98; H, 6.03.

Methyl 6-O-(p-coumaroyl)-β-D-glucopyranoside (1h).—The appropriate fractions from the ion-exchange column were processed to afford 1h as a white foam (425 mg, 94%): $[\alpha]_D$ – 18.1° (c 1.0, acetone– H_2O); NMR (9:1 acetone- d_6 – D_2O): δ_H 3.20 (dd, 1 H, $J_{2,1}$ 7.8 Hz, $J_{2,3}$ 9.0 Hz, H-2); 3.35–3.47 (m, 2 H, H-3, 4); 3.41 (OCH₃); 3.55 (m, 1 H, H-5); 4.22 (d, 1 H, H-1); 4.28 (dd, 1 H, $J_{6a,5}$ 6.1 Hz, J_{gem} 11.9 Hz, H-6a); 4.47 (dd, 1 H, $J_{6b,5}$ 2.1 Hz, H-6b); 6.34 (d, 1 H, $J_{8',7'}$ 16.0 Hz, H-8'); 6.85 (m, 2 H, H-3', 5'); 7.49 (m, 2 H, H-2', 6'), 7.60 (d, 1 H, H-7'); δ_C (acetone- d_6), 56.7 (OCH₃).

Anal. Calcd for $C_{16}H_{20}O_8$: C, 56.47; H, 5.92. Found: C, 56.30; H, 6.02.

Methyl 6-O-feruloyl-β-D-glucopyranoside (1i).—Processing of the appropriate fractions from the ion-exchange column gave 1i as a white foam (392 mg, 90%): $[\alpha]_D$ –16.9° (c 1.47, acetone–H₂O); NMR (9:1 acetone-d₆–D₂O): δ_H 3.21 (dd, 1 H, $J_{2,1}$ 7.8 Hz, $J_{2,3}$ 9.0 Hz, H-2); 3.36–3.49 (m, 5 H, OCH₃, H-3, 4); 3.56 (ddd, 1 H, $J_{5,6b}$ 2.1, $J_{5,6a}$ 6.0 Hz, $J_{5,4}$ 9.5 Hz, H-5); 3.85 (OCH₃); 4.23 (d, 1 H, H-1); 4.30 (dd, 1 H, J_{gem} 12.0 Hz, H-6a); 4.47 (dd, 1 H, H-6b); 6.39 (d, 1 H, $J_{8',7'}$ 15.9 Hz, H-8'); 6.84 (d, 1 H, $J_{5',6'}$ 8.2 Hz, H-5'); 7.08 (dd, 1 H, $J_{6',2'}$ 2.0 Hz, H-6'); 7.26 (d, 1 H, H-2'); 7.59 (d, 1 H, H-7'); δ_C (acetone- J_6), 56.3, 56.7 (OCH₃).

Anal. Calcd for C₁₇H₂₂O₉: C, 55.13; H, 5.99. Found: C, 55.03; H, 6.05.

Acetylations.—The crystalline 4'-acetoxycinnamates (1f and g, and 2f and g; 200 mg) were dissolved in pyridine (2 mL) and acetic anhydride (1.5 mL) and left overnight. The reaction was quenched with 95% EtOH, diluted with toluene, and evaporated to a syrup. The syrup was diluted with CH₂Cl₂ and washed successively with 3% HCl, H₂O, and aq NaHCO₃. Standard processing gave the fully acetylated materials described below.

Methyl 2,3,4-tri-O-acetyl-6-O-(4'-O-acetyl-p-coumaroyl)-β-D-glucopyranoside (1j).

Compound 1j crystallized from 95% EtOH as small white needles: mp 158.5–160°; $[\alpha]_{\rm D}$ 11.5° (c 1.86, acetone); NMR (acetone- d_6): $\delta_{\rm H}$ 1.93, 1.98, 2.00 and 2.27 (Ac), 3.46 (OCH₃), 4.01 (dt, 1 H, $J_{5,4}$ 10.0 Hz, $J_{5,6a}+J_{5,6b}$ 7.6 Hz, H-5), 4.34 (m, 2 H, H-6a, 6b), 4.67 (d, 1 H, $J_{1,2}$ 8.0 Hz, H-1), 4.92 (dd, 1 H, $J_{2,3}$ 9.7 Hz, H-2), 5.09 (t, 1 H, $J_{4,3}+J_{4,5}$ 19.4 Hz, H-4), 5.26 (t, 1 H, $J_{3,2}+J_{3,4}$ 19.1 Hz, H-3), 6.56 (d, 1 H, $J_{7',8'}$ 16.0 Hz, H-7'), 7.20 (m, 2 H, H-3',5'), 7.71 (d, 1 H, H-8'), 7.74 (m, 2 H, H-2',6'); $\delta_{\rm C}$ 20.5, 20.5, 20.6 and 20.9 (Ac), 56.8 (OCH₃), 169.4, 169.6, 170.0, 170.3 (C=O).

Anal. Calcd for C₂₄H₂₈O₁₂: C, 56.69; H, 5.55. Found: C, 56.71; H, 5.57.

Methyl 2,3,4-tri-O-acetyl-6-O-(4'-O-acetylferuloyl)-β-D-glucopyranoside (1k). Compound 1k crystallized from 95% EtOH as small white needles: mp 152–153°; $[\alpha]_D$ 12.6° (c 1.51, acetone); NMR (acetone- d_6): δ_H 1.94, 1.99, 2.00 and 2.25 (Ac); 3.46 and 3.89 (OCH₃); 4.00 (m, 1 H, $J_{5,4}$ 10.0 Hz, H-5), 4.32 (m, 1 H, $J_{6a,5}$ 5.0 Hz, J_{gem} 12.3 Hz, H-6a), 4.36 (m, 1 H, $J_{6b,5}$ 2.5 Hz, H-6b), 4.66 (d, 1 H, $J_{1,2}$ 8.0 Hz, H-1), 4.92 (dd, 1 H, $J_{2,3}$ 9.7 Hz, H-2), 5.09 (t, 1 H, $J_{4,3}$ + $J_{4,5}$ 19.4 Hz, H-4), 5.26 (t, 1 H, $J_{3,2}$ + $J_{3,4}$ 19.0 Hz, H-3), 6.60 (d, 1 H, $J_{8',7'}$ 16.0 Hz, H-8'), 7.11 (d, 1 H, $J_{5',6'}$ 8.1 Hz, H-5'), 7.25 (dd, 1 H, $J_{2',6'}$ 1.83 Hz, H-6'), 7.47 (d, 1 H, H-2'), 7.68 (d, 1 H, H-7'); δ_C 20.4, 20.5, 20.5 and 20.6 (Ac); 56.4 and 56.8 (OCH₃); 168.8, 169.6, 169.9 and 170.3 (C=O).

Anal. Calcd for C₂₅H₃₀O₁₃: C, 55.76; H 5.62. Found: C, 55.84; H, 5.61.

Methyl 2,3,4-tri-O-acetyl-6-O-(4'-O-acetyl-p-coumaroyl)-β-D-galactopyranoside (2j). Compound 2j was isolated as a clear syrup: $[\alpha]_D$ –27.4° (c 2.23, acetone); NMR: δ_H (CDCl₃) 1.95, 2.03, 2.13 and 2.27 (Ac), 3.49 (OCH₃), 3.96 (t, 1 H, $J_{5,6a}$ + $J_{5,6b}$ 14.3 Hz, H-5), 4.18 (dd, 1 H, $J_{6a,5}$ 6.9 Hz, J_{gem} 11.3 Hz, H-6a), 4.36 (dd, 1 H, $J_{6b,5}$ 6.5 Hz, H-6b), 4.39 (d, 1 H, $J_{1,2}$ 7.3 Hz, H-1), 5.01 (dd, 1 H, $J_{3,4}$ 3.4 Hz, $J_{3,2}$ 10.5 Hz, H-3), 5.19 (dd, 1 H, H-2), 5.43 (d, 1 H, H-4), 6.32 (d, 1 H, $J_{7',8'}$ 16.0 Hz, H-8'), 7.09 (d, 2 H, J 8.6 Hz, H-3',5'), 7.50 (d, 2 H, H-2',6'), 7.63 (d, 1 H, H-7'); δ_C (acetone- d_6) 20.5, 20.5, 20.6, 20.9 (Ac), 56.7 (OCH₃), 169.4, 169.7, 170.2, 170.7 (C=O).

Anal. Calcd for C₂₄H₂₈O₁₂: C, 56.69; H, 5.55. Found: C, 56.61; H, 5.59.

Methyl 2,3,4-tri-O-acetyl-6-O-(4'-O-acetylferuloyl)-β-D-galactopyranoside (2k). Compound 2k was isolated as a clear syrup: $[\alpha]_D$ – 27.0 (c 3.68, acetone); NMR (acetone- d_6): δ_H (CDCl₃) 1.95, 2.03, 2.14 and 2.28 (Ac); 3.49 and 3.83 (OCH₃); 3.95 (dt, 1 H, $J_{5,4}$ 1.1 Hz, $J_{5,6a}$ + $J_{5,6b}$ 13.4 Hz, H-5), 4.18 (dd, 1 H, $J_{6a,5}$ 7.1 Hz, J_{gem} 11.2 Hz, H-6a), 4.37 (dd, 1 H, $J_{6b,5}$ 6.4 Hz, H-6b), 4.40 (d, 1 H, $J_{1,2}$ 7.9 Hz, H-1), 5.01 (dd, 1 H, $J_{3,4}$ 3.4 Hz, $J_{3,2}$ 10.5 Hz, H-3), 5.19 (dd, 1 H, H-2), 5.43 (dd, 1 H, H-4), 6.32 (d, 1 H, $J_{8',7'}$ 16.0 Hz, H-8'), 7.00–7.09 (m, 3 H, H-2', 5', 6'), 7.61 (d, 1 H, H-7'); δ_C 20.4, 20.4, 20.5 and 20.6 (Ac), 56.3 (Ar-OCH₃), 56.7 (OCH₃), 168.8, 169.7, 170.2 and 170.7 (C=O).

Anal. Calcd for C₂₅H₃₀O₁₃: C, 55.76; H, 5.62. Found: C, 55.81; H, 5.64.

Procedure for TBDMS-acylation sequence.—The generation of methyl 2,3-di-O-acetyl-5-O-tert-butyldimethylsilyl- α -L-arabinofuranoside (3c) was followed by O-desilylation and subsequent coupling to the 4'-acetoxycinnamoyl chlorides. The peracetylated esters 3g and 3h were deacetylated in pyrrolidine-95% EtOH.

Methyl 2,3-di-O-acetyl-5-O-tert-butyldimethylsilyl-α-L-arabinofuranoside (3c). Methyl α-L-arabinofuranoside 21,26 (2.54 g, 15.5 mmol) was dissolved in pyridine (20 mL) under an N₂ atmosphere, and cooled in an ice-water bath. tert-Butyl-dimethylsilyl chloride (Petrarch, 2.44 g, 16.2 mmol) was added, the mixture was stirred for 5 min, and the flask was removed from the bath and left stirring overnight. Purification of the reaction mixture (silica gel chromatography) provided methyl 5-O-tert-butyldimethylsilyl-α-L-arabinofuranoside (3b) as a clear syrup in 89% yield: $[\alpha]_D - 79.7^\circ$ (c 2.68, acetone); NMR (acetone- d_6): δ_C 3.2 [Si(CH₃)₂], 18.8 [C(CH₃)], 26.2 [C(CH₃)₃], 54.7 (OCH₃), 64.2 (C-5), 78.5 (C-3), 82.5 (C-4), 85.5 (C-2), 110.1 (C-1).

Anal. Calcd for C₁₂H₂₆O₅Si: C, 51.77; H 9.41. Found: C, 51.75; H, 9.44.

The crude reaction mixture containing **3b** could be acetylated without purification, providing a one-pot procedure for the formation of **3c**. Acetic anhydride (6 mL, 4.1 equiv) was added to the crude solution of **3b**, and the mixture stirred for 5 h. The reaction mixture was quenched with 95% EtOH and diluted with CH_2Cl_2 . The organic phase was washed subsequently with H_2O , 3% aq HCl, H_2O , and aq $NaHCO_3$. The organic layer was dried, filtered, and evaporated to a syrup. Two additions and evaporations of toluene eliminated the remaining pyridine. Purification by silica gel chromatography (40 g, solvent *A*) gave **3c** as a clear syrup (4.73 g, 84.3% from **3a**): $[\alpha]_D - 66.1^\circ$ (*c*, 1.08, acetone); NMR (acetone- d_6): δ_H 0.090, 0.096 $[Si(CH_3)_2]$; 0.92 $[C(CH_3)_3]$; 2.03, 2.04 (Ac); 3.33 (OCH_3) ; 3.84, 3.87 (s's, 2 × 1 H, H-5a, 5b); 4.04 (m, 1 H, H-4); 4.86 (bs, 1 H, H-1); 4.97 (bd, 1 H, $J_{2,3}$ 1.7 Hz, H-2); 5.08 (dd, 1 H, $J_{3,4}$ 5.1 Hz, H-3); δ_C 3.5 $[Si(CH_3)_2]$; 18.8 $[C(CH_3)]$; 20.7 (Ac); 26.2 $[C(CH_3)_3]$; 54.5 (OCH_3) ; 63.2 (C-5); 77.7 (C-3); 82.7 (C-4); 83.5 (C-2); 107.5 (C-1); 170.3, 170.3 (Ac).

Anal. Calcd for C₁₆H₃₀O₇Si: C, 53.01; H, 8.34. Found: C, 52.91; H, 8.48.

Methyl 2,3-di-O-acetyl-α-L-arabinofuranoside (3d). Syrupy 3c (4.01 g, 11.05 mmol) was diluted with 80% HOAc (20 mL) and left for 24 h at room temperature. The product was evaporated to a syrup and diluted with CH₂Cl₂, which was subsequently washed twice with H₂O and once with aq NaHCO₃. Standard work-up gave 3d (2.50 g, 91%) as a clear syrup: $[\alpha]_D$ –91.9° (c 1.18, acetone); NMR (CDCl₃): δ_H 2.06, 2.07 (Ac); 3.36 (OCH₃); 3.73–3.89 (m, 2 H, H-5a, 5b); 4.06 (m, 1 H, H-4); 4.88 (bs, 1 H, H-1); 4.99 (dd, 1 H, $J_{3,2}$ 1.6 Hz, $J_{3,4}$ 5.3 Hz, H-3); 5.05 (d, 1 H, H-2); δ_C 20.6, 20.7 (Ac); 54.7 (OCH₃); 61.9 (C-5), 77.1 (C-3); 81.6 (C-4); 82.8 (C-2); 106.5 (C-1); 169.7, 170.5 (C=O).

Anal. Calcd for $C_{10}H_{16}O_7$: C, 48.39; H, 6.50. Found: C, 48.26; H, 6.53.

Crude 3c, obtained without silica gel chromatography, can also be subjected to the O-desilylation reaction to provide 3d in 70% overall yield (based on 3a).

Methyl 2,3-di-O-acetyl-5-O-(4'-O-acetyl p-coumaroyl)- α -L-arabinofuranoside (3e). A mixture of 3d (818 mg, 3.3 mmol) and 4'-acetoxy-p-coumaroyl chloride (806 mg, 3.59 mmol) in pyridine (5 mL) was stirred for 5 h. The mixture was diluted with toluene and evaporated to a syrup. Purification by silica gel chromatography (40 g, solvent B) gave 3e (1.20 g, 84%) as a clear viscous syrup: $[\alpha]_D$ -65.4° (c 2.68)

acetone); NMR (acetone- d_6): $\delta_{\rm H}$ 2.06 (Ac), 2.26 (Ac), 3.35 (OCH $_3$), 4.30 (dt, 1 H, $J_{4,5a}$ 3.3 Hz, $J_{4,3}+J_{4,5b}$ 10.7 Hz, H-4), 4.37 (dd, 1 H, $J_{5a,4}$ 5.5 Hz, J_{gem} 11.8 Hz, H-5a), 4.55 (dd, 1 H, H-5b), 4.94 (s, 1 H, H-1), 5.03 (d, 1 H, $J_{2,3}$ 1.6 Hz, H-2), 5.07 (ddd, 1 H, $J_{3,1}$ 0.7 Hz, $J_{3,4}$ 5.2 Hz, H-3), 6.55 (d, 1 H, $J_{8',7'}$ 16.1 Hz, H-8'), 7.18–7.20 (m, 2 H, H-3', 5'), 7.70–7.74 (m, 3 H, H-2', 6', 7'); $\delta_{\rm C}$ 20.6, 20.7, 20.9 (Ac); 54.8 (OCH $_3$); 169.4, 170.1, 170.6, (Ac).

Anal. Calcd for C₂₁H₂₄O₁₀: C, 57.8; H, 5.54. Found: C, 58.06; H, 5.60.

Methyl 2,3-di-O-acetyl-5-O-(4'-O-acetylferuloyl)-α-1-arabinofuranoside (3f). Compound 3f was prepared as described for 3e in 89% yield: $[\alpha]_D$ –53.5° (c 1.53, acetone); NMR (acetone- d_6): δ_H 2.06, 2.07, 2.24 (OAc); 3.35, 3.89 (OCH $_3$); 4.28 (m, 1 H, H-4); 4.34 (dd, 1 H, $J_{5a,4}$ 5.5 Hz, J_{gem} 11.8 Hz, H-5a); 4.54 (dd, 1 H, $J_{5b,4}$ 3.2 Hz, H-5b); 4.93 (bs, 1 H, H-1); 5.02 (bd, 1 H, $J_{2,3}$ 1.6 Hz, H-2); 5.05 (m, 1 H, $J_{3,4}$ 5.2 Hz, H-3); 6.60 (d, 1 H, $J_{8',7'}$ 16.0 Hz, H-8'); 7.11 (d, 1 H, $J_{5',6'}$ 8.1 Hz, H-5'); 7.27 (dd, 1 H, $J_{6',2'}$ 1.8 Hz, H-6'); 7.48 (d, 1 H, H-2'); 7.70 (d, 1 H, H-7'); δ_C 20.4, 20.6, 20.7 (Ac); 54.8, 56.3 (OCH $_3$); 168.8, 170.1, 170.6 (Ac).

Anal. Calcd for C₂₂H₂₆O₁₁: C, 56.65; H, 5.62. Found: C, 56.68; H, 5.69.

Methyl 5-O-(p-coumaroyl)-α-L-arabinofuranoside (3g). Syrupy 3e (250 mg) was dissolved in 95% EtOH (5 mL). Pyrrolidine (500 μL) was added, and the mixture was left without stirring for 48 h. Neutralization and purification as described for the piperidine deacetylation reactions afforded 3g (83%) as a clear syrup: $[\alpha]_D$ – 89.0° (c 1.0, acetone); NMR (acetone- d_6): δ_H 3.31 (OCH₃), 3.96 (dd, 1 H, $J_{3,2}$ 3.7 Hz, $J_{3,4}$ 6.4 Hz, H-3), 4.06 (dd, 1 H, $J_{2,1}$ 1.6 Hz, H-2), 4.14 (dt, 1 H, $J_{4,5b}$ 3.6 Hz, $J_{4,5a}$ + $J_{4,3}$ 12.5 Hz, H-4), 4.27 (dd, 1 H, J_{gem} 11.8 Hz, H-5a), 4.40 (dd, 1 H, H-5b), 4.80 (d, 1 H, H-1), 6.36 (d, 1 H, $J_{8',7'}$ 16.0 Hz, H-8'), 6.87–6.90 (m, 2 H, H-3', 5'), 7.51–7.54 (m, 2 H, H-2', 6'), 7.64 (d, 1 H, H-7'); δ_C 55.0 (OCH₃).

Anal. Calcd for C₁₅H₁₈O₇; C, 58.06; H, 5.85. Found: C, 57.93; H, 5.75.

Methyl 5-O-feruloyl- α -L-arabinofuranoside (3h). Compound 3h was prepared as described for 3g in 73% yield as a hygroscopic solid which can be crystallized from CH₂Cl₂ (ref. 21): mp 71–72°; $[\alpha]_D$ –73.8° (c 0.70, acetone).

Anal. Calcd for C₁₆H₂₀O₈: C, 56.47; H 5.92. Found: C, 56.38; H, 5.79.

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